

# Ethanol, UV method

## Alternative Procedures

### Introduction

This kit has been developed to work in cuvettes with a standard pathlength of 1 cm, as described in the respective “Product Brochure”. However, it can be adapted for use in 96-well microplates or in auto-analysers (micro-volume formats) with minimal assay optimisation. Basically, the assay volumes for the cuvette format have to be reduced approximately 10-fold for use in microplate format or in auto-analyser format. However, when using these micro-volume formats, you must be aware that the radiation pathlength is usually smaller than the standard cuvette pathlength of 1 cm. Thus, to perform the calculation of the amount of analyte in the samples under analysis follow one of the three strategies described in the section below.

### Strategies for analyte calculation

Auto-analysers use reaction volumes of approximately 0.315 ml and pathlengths from 4 to 8 mm, which is similar to a standard 96-well microplate in which the same reaction volume would have a pathlength of 6 or 7 mm (similar assay volumes). Therefore, in both formats (96-well microplate and auto-analysers systems), the calculation of the analyte must be done by one of the three possible methods described below:

#### 1. Using the pathlength conversion factor

This is the easiest method to perform the calculation of the analyte. However, it requires a microplate reader with pathlength conversion capacity, i.e., the apparatus can detect the pathlength of each well and convert the individual readings to a 1 cm pathlength (cuvette format). In the case of auto-analysers, the absorbance readings should be directly converted to a 1 cm pathlength. This will allow the calculation of the analyte content as described in the “Product Brochure”, provided with the kit and available at the NZYTech website.

#### 2. Using one standard curve

In this method, it is necessary to perform a standard curve of the analyte on each microplate that contains the test samples, or in the auto-analyser, and calculate the result from the standard curve of analyte concentration vs. absorbance. The standard curve can be performed by using the control solution provided in the kit.

#### 3. Using two standard curves

The most complicated method is to perform standard curves of the analyte in both the cuvette format (i.e. with a 1 cm of radiation pathlength) and the 96-well microplate or auto-analyser formats, and use these results to obtain a mean conversion factor between the cuvette procedure values and the alternative procedure values.

The standard curves can be performed by using the control solution provided in the kit.

### Procedure (microplate)

Wavelength: 340 nm

Microplate: 96-well (e.g. clear flat-bottomed, glass or plastic)

Temperature: ~ 25°C

Final volume: 0.254 ml

Sample solution: 0.025-1.2 µg of ethanol per cuvette (in 0.010-0.200 ml sample volume)

Pipette into cuvettes (mL)	Blank	Sample	Standard
Distilled water	0.210	0.200	0.200
Sample	-	0.010	-
Standard solution	-	-	0.010
Solution 1 (PP <sub>i</sub> buffer)	0.020	0.020	0.020
Solution 2 (NAD <sup>+</sup> )	0.020	0.020	0.020
Suspension 3 (Al-DH)	0.002	0.002	0.002
Mix, measure the absorbance of the solutions (A1) after ~2 min and start the reaction by addition of			
Suspension 4 (ADH)	0.002	0.002	0.002
Mix, measure the absorbance of the solutions (A2) at the end of the reaction (approx. 5 min)*			

Mixture can be obtained using microplate shaker, shake function on a microplate reader, or repeated aspiration (e.g. using a pipettor set at 50 - 100 µL volume).

\* if necessary, continue to read the absorbances at 1 min intervals until the reaction ends.

### Calculation (microplate procedure)

$$g/L = \frac{\Delta A_{\text{sample}}}{\Delta A_{\text{standard}}} \times g/L \text{ standard} \times F$$

If the sample is diluted during preparation, the result must be multiplied by the dilution factor, F.



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